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Chemical Reactions Answer

Key. ...  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$

20 5)  $1 \text{Pb}(\text{OH})_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O}$

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Section 2 Types of Bonds.

Gain or loss of electrons.

Atoms gain or lose electrons  
to become stable. Atoms that

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Bonds or lose electrons are called ions. It is the electric forces between oppositely charged particles, such as ions, that hold compounds together. Bond Formation.

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boiling point =  $100^{\circ}\text{C}$  2.

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1. atomic number 1. bubble  
chamber 2. element 2.  
particle accelerator 3.  
chemical symbol 3.

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1. ionic bond. 2. ion. 3.



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chemical bond. 4. compound.  
5. covalent bond. 6.  
molecule. 7. formula. 8.  
electron dot diagram. 9.  
polar bond. 10. polar  
molecule ... electrons are  
shared equally by atoms. In  
a polar bond, electrons are

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Bonds unequally. (3/2)

Reinforcement. Section 1

(page 81) 1. the formation  
of new substances that have.  
properties ...

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midst of them is this

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atom B gains these  
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REACTIONS Reinforcement KEY  
CONCEPT Life depends on  
chemical reactions. At the

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most fundamental level,  
every process that takes  
place in an organism depends  
on chemical reactions. In a  
chemical reaction,  
substances are changed into  
different substances by the  
breaking and forming of

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