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Sodium oxide and potassium oxide dissolve in water to produce alkalis: • $\text{Na}_2\text{O}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow 2\text{NaOH}(\text{aq})$ • $\text{K}_2\text{O}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow 2\text{KOH}(\text{aq})$ • $2\text{K}(\text{s}) + 2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{KOH}(\text{aq}) + \text{H}_2 \dots$

~~CBSE Class 10th Science Board Exam: Important Chemical Reactions~~

On April 14, the CBSE had announced the cancellation of class 10 exams and the postponement of class 12 exams in view of the surge in coronavirus cases. For breaking news and live news updates ...

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