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by. dvann123. Atoms, Law of Conservation of mass & proportions, counting atoms related terms. Terms in

this set (28) law of conservation of mass. states that mass is neither created nor destroyed during

ordinary reactions or physical changes.

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destroyed nor created during ordinary chemical reactions or physical changes. law of definite

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Modern Chemistry Chapter 3 Notes Modern Chemistry Chapter 3. law of conservation of mass. law of

definite proportions. law of multiple proportions. atom. States that matter can be neither destroyed

nor created during... The fact that a chemical compounds contains the same elements... if two or more

different compounds are composed of the same ...

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Modern Chemistry Chapter 3. Atoms: The Building Blocks of Matter. law of conservation of mass- mass is

neither created nor destroyed during ordinary chemical reactions or physical changes. e.g. 20 g A + 20

g B 40 g AB.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following

questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative

electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone

or in molecules containing the

3 Atoms: The Building Blocks of Matter

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Unformatted text preview: Modern Chemistry Chapter 3 Atoms: The Building Blocks of Matter law of conservation of mass mass is neither created nor destroyed during ordinary chemical reactions or physical changes e.g. $20\text{ g A} + 20\text{ g B} \rightarrow 40\text{ g AB}$ law of definite proportions- a chemical compound contains the same elements in exactly the same proportions by mass regardless of the size of the sample or the source of the compound e.g. If 10 grams of A combine with 20 grams of B to form compound AB, how ...

Modern Chemistry Chapter 3 - Modern Chemistry Chapter 3 ...

CHAPTER 3 TEST Class Atoms: The Building Blocks of Matter MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice best completes the statement or answers the question. The behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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