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C H C H C H O O O CO CO CO H O H O H O number of
moles of unknown substance O CO H O C H CO H O C
H O H O C H O CO use mole ratio 5 mol O 2 1 mol C 3
H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...

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MASTER 4 ... 94 Chemistry: Matter and Change •
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Calculating Atomic Mass MATH SKILLS

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MATH SKILLS TRANSPARENCY MASTER. Use with Chapter 4, Section 4.3. 4. Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used medically to treat some mental disorders. Analyze the Problem. You are given the data in the table.

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MASTER 4 ...

Chapter 11, 1.2 MATH SKILLS TRANSPARENCY
WORKSHEET 1. Explain what a crystalline solid is. 18
Use with Chapter 12, Section 2.3 1. The reaction of
iron(III) oxide (Fe_2O_3) and hydrogen (H_2) is
represented by the following unbalanced chemical
equation. $\text{Fe}_2\text{O}_3(\text{s}) + \text{H}_2(\text{g}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l})$
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Chapter 2: Analyzing Data

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MATH SKILLS TRANSPARENCY WORKSHEET

Determining Numbers of Ions Use with Chapter 7,
Section 7.3 1. Write a simple word equation that
illustrates what must be true of total positive charge

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and total negative charge in an ionic compound. 2.
How many potassium ions (group 1) would be related
to balance the charge of each of the following in a
compound?

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Chapter 4 Math Skills Transparency Masters

Calculating Atomic Mass MATH SKILLS

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Math Skills Transparency Answers

82 Chemistry: Matter and Change • Chapter 25 Math Skills Transparency Worksheets 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 2. Explain how both mass number and atomic number have been conserved in equation 1. 3. Write the nuclear symbol for the missing term in equation 2. 4.

balancing nuclear equations

Math Skills Transparency Worksheets Chemistry: Matter and Change • Chapter 5 13 1. Look at the two waves shown. What is the speed of each wave? 2. Look at the two waves shown. Which wave has a

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higher frequency? Which wave has a longer wavelength? 3. Assume that wave A has a wavelength of 699 nm. Calculate the frequency of the wave. Show your work. 4.

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Chapter 25, Section 25.3 42 82 1. In equation 1, what
do the numbers 238 and 92, written to the left of the
symbol U, represent? 238 is the mass number of the
isotope, and 92 is the atomic number of the isotope.
2. Explain how both mass number and atomic number
have been conserved in equation 1.

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of Ions Use with Chapter 7, Section 73 1 Write a

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simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound 2 How many potassium ions (group 1)

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