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Electroplating - Wikipedia

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Problem 2.19 1/1 Determine the solubility product K_{sp} for $PbSO_4$. The desired
equilibrium is $PbSO_4 \rightleftharpoons Pb^{2+} + SO_4^{2-}$ We can write this as the sum of two
electrochemical equations $PbSO_4 + 2e^- \rightleftharpoons Pb + SO_4^{2-}$ $Pb \rightleftharpoons Pb^{2+} + 2e^-$ =

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