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Question 1. Solubility is the amount of a substance that will dissolve a given amount of

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NEXT PAGE. -6-(c) Calculate the
concentration of urea, in mol/L, in
the saturated solution at 20.°C.

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(d) The student also determines that the concentration of urea in a saturated solution at 25°C is $19.8 \text{ g} \cdot \text{M}$. Based on this information, is the dissolution of urea endothermic or exothermic?

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measure heat change is the calorimeter. Two popular types of calorimeters are the coffee cup calorimeter and bomb calorimeter.

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