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Chemistry Project 2

What is would be the

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molality of a solution
obtained by mixing equal
volumes of 30% by weight
`H...

Two metallic oxides contain
`27.6%` and `30%` oxygen,
respectively. If the formula
of the *Energy Levels, Energy*

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11. In your own words, what

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is the difference between a sublevel and an orbital?

Orbitals are contained within sublevels. A sublevel is a grouping of orbitals.

12. How many electrons can fit in each of the following energy levels: 1st energy

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level = 2 because only 2 can fit in an s sublevel and the first energy level only has an s sublevel.

Information: Energy Levels and Sublevels

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Page 143 and 144: 143 9.
Consider your answers to
que. Page 145 and 146: 145
Critical Thinking Questions
5. Page 147 and 148: 147 8.
If a substance is a
Bronsted. Page 149 and 150:
149 6. Use the information

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given an. Page 151 and 152:

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Page 153 and 154: 153 11.

Find the pH of a 0.85 M sol

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11. The energy difference
between the 1st and 2nd

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energy levels in a certain atom is 5.01×10^{-19} J. What frequency of light is necessary to excite an electron in the 1st energy level? 12. a) What is the frequency of light given off by an electron that loses

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4.05 x 10⁻¹⁹ J of energy as it moves from the 2nd to the 1st energy level?

*Critical Thinking Questions:
Bohr's reasoning*

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