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 $4\text{NH}_3(\text{g}) + 5\text{O}_2(\text{g}) \rightleftharpoons 4\text{NO}(\text{g}) + 6\text{H}_2\text{O}(\text{g})$ $\Delta H = -1264 \text{ kJ mol}^{-1}$
Describe the effect, if any, of each of the following changes on the equilibrium concentration of nitrogen monoxide in a particular equilibrium mixture, giving a reason in each case. (i) Increasing the pressure, at

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constant temperature [2]

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– 2 – Section A Answer all questions.
Write your answers in the boxes provided.

1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC} - \text{COOH} \cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of $\text{NaOH}(\text{aq})$. 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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Ethane, C_2H_6 , and disilane, Si_2H_6 , are both hydrides of group 4 elements with similar structures but with different chemical properties. (a) (i) Deduce the Lewis (electron dot) structure for Si_2H_6

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showing all valence electrons. [1] (ii) State and explain the H – Si – H bond angle in Si₂H₆. [2]

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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle.

C ₂ H ₄ (g)	+ H ₂ (g)	C ₂ H ₆ (g)	3 O ₂	2 CO ₂ (g)	+ 3H ₂ O
– 141	1	1		– 1	– 286
kJ mol ⁻¹					
– 1	– 1560	kJ mol ⁻¹			

(hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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