

## Chemistry Matter Change Chapter 15 Answer Key

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Chemistry: Matter and Change Chapter 15-Energy and Chemical Change. STUDY. PLAY. energy. the capacity to do work or produce heat. law of conservation of energy. states that in any chemical reaction or physical process, energy can be converted from one form to another, but it neither created nor destroyed.

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*Calculate the heat required to melt 25.7 g of solid ...*

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*A 155-g sample of an unknown substance was heated from ...*

Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe<sub>2</sub>O<sub>3</sub>(s) is -851.1 kJ, whether the reaction occurs in a single step (ΔH<sub>4</sub>, shown on the left) or in three hypothetical steps (shown on the right) that involve ...

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