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Geometry Ionic

Radius Ionic vs.

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vs. Nonpolar FSc

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~~theory 11th~~

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~~Theories of~~

~~Covalent Bonding~~

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CHAPTER 6 REVIEW

Chemical Bonding

SECTION 1 SHORT

ANSWER Answer

the following
questions in the
space provided.

1. a A chemical
bond between
atoms results
from the
attraction
between the
valence

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electrons and of
different atoms.

(a) nuclei (c)
isotopes (b)

inner electrons

(d) Lewis
structures 2. b

A covalent bond
consists of (a)
a shared
electron.

6 Chemical
Bonding -

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Effingham County

School District

Chapter 6:

Chemical Bonding

Section 1-

Introduction to

Chemical Bonding

Objectives:

define chemical

bond;

differentiate

between covalent

and ionic

bonding; explain

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why bonding
occurs; use
the...

Section 3

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Chemical Bonding

section 1

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and Molecular

Compounds

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Ionic Compounds

section 4

Metallic Bonding

section 5

Molecular

Geometry CHAPTER

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Chemical Bonding

Key Terms

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Chemical bond no
npolar-covalent
bond . . .

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Chemical Bonding
A chemical bond

is a mutual
electrical
attraction
between the
nuclei and
valence
electrons of

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different atoms that binds the atoms together. When atoms form a chemical bond, their valence electrons are redistributed to make the atoms more stable. The way the electrons are redistributed determines the

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type of bond.

Chemical bonding
that results
from the

electrical

Answers

CHAPTER 6

Chemical Bonding

Modern Chemistry

1 Chemical

Bonding CHAPTER

6 Chemical

Bonding SECTION

1 Introduction

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to Chemical

Bonding

OBJECTIVES 1.

Define Chemical

bond. 2. Explain

why most atoms

form chemical

bonds. 3.

Describe ionic

and covalent

bonding.. 4.

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Chemical Bonding

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Chemical Bonding

Section 3
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SECTION 3 SHORT

ANSWER Answer

the following

questions in the

space provided.

1. a The

notation for

sodium chloride,

NaCl, stands for

one (a) formula

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unit. (c)

crystal. (b)

molecule. (d)

atom. 2. d In a

crystal of an

ionic compound,

each cation is

surrounded by a

number of (a)

molecules. (c)

dipoles. (b)

positive ions.

(d) negative

ions.

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6 Chemical

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Somerset Canyons

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Jennie L.

Borders.

Standards. SPS1.

Students will

investigate our

current

understanding of

the atom. b.

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Compare and contrast ionic and covalent bonds in terms of electron movement. SPS2. Students will explore the nature of matter, its classification and its system for naming types of matter.

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to Chemical

Bonding Atoms

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independent
particles in
nature. Most

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Chemical Bonding

DRAFT. ... Two

or more atoms

held together by

a chemical bond

is called what?

answer choices .

A molecule. A

molecure. Tags:

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Question 37 .

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terms, and more

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CHAPTER 6 REVIEW

Chemical Bonding

SECTION 3 SHORT

ANSWER Answer

the following

questions in the
space provided 1

a The notation
for sodium

chloride, NaCl ,
stands for one

- (a) formula unit
- (c) crystal (b)
- molecule (d)

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atom 2 d In a
crystal of an
ionic compound,
each cation is

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Answers

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Section 2

Covalent Answer

Key ...

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BONDING SECTION

3 SHORT ANSWER

ANSWER THE

FOLLOWING

QUESTIONS IN THE

SPACE PROVIDED 1

A THE NOTATION

FOR SODIUM

CHLORIDE NaCl

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Covalent Bonding

Questions for

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library or

borrowing from

your friends to

entre them. This

is an agreed

simple means to

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lead by on-line.

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chemistry

chapter 6

section 3, 4, 5.

ionic compound.

formula unit.

lattice energy.

polyatomic ion.

A compound that

consists of

positive and

negative ions.

the lowest whole-

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Chapter 6

number ratio of
ions in an ionic
compound. Page
3/10

Worksheet

Answers

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