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Thermal

Heat. the transfer of thermal energy from
one object to another because of a

difference in temperature. Temperature. a

measure of how hot or cold an object is

compared to a reference point. Absolute

Zero. a temperate of 0 kelvins. Thermal

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Expansion: the increase in volume of a material when its temperature increases.

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20 terms from Pearson's Physical Science
Chapter 16. Terms in this set (20) Heat.

The transfer of thermal energy from one object to another because of a temperature difference. Temperature. The measure of how hot or cold something is compared to a reference point, related to average kinetic energy. Absolute Zero.

Pearson Physical Science Workbook

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Heat. the transfer of thermal energy from one object to another as the result of a difference in temperature. True. T/F: On the Celsius Scale, the reference points for temperature are the freezing and boiling

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points of water. thermal energy... Chapter
16: Thermal Energy and Page 9/30
And Heat

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Chapter 16 – Thermal Energy and Heat
Section 16.1 – Thermal Energy and
Matter In the 1700 ' s most scientists
thought that heat was a fluid called caloric
that flowed between objects. Chapter 16
– Thermal Energy and Heat

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Heat 16.1 – Thermal Energy and Matter
. Work and Heat Heat is the transfer of
thermal energy from one object to another
because of temperature differences Heat
flows spontaneously from hot objects to
cold objects Imagine two glasses with

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thermal energy and chapter 16 thermal
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Matter • Heat – transfer of thermal
energy from one object to another because
of temp. difference • Heat flows

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