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~~11 BASIC Carbon Compound Study Guide Answer~~

SAMPLE ANSWER: Organisms use carbon compounds to form four types of molecules: lipids, carbohydrates, nucleic acids, and proteins. SAMPLE ANSWER: A lot of what that happens in an organism is based on chemical reactions. SAMPLE

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ANSWER: Enzymes are proteins that speed up chemical reactions that take place in cells.

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Carbon Compound Study Guide Answer

Covalently bonded compound that contains carbon, excluding carbonates and

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oxides. Functional Group Portion of a molecule that is active in a chemical reaction and that determines the properties of many organic compounds. Biology Section 3-1 Review: Carbon Compounds Flashcards ... Carbon Compound Study Guide Answer Key Carbon Compound Study Guide

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~~Carbon Compound Study Guide Answer
Key~~

Answer to: The empirical formula of a compound containing carbon, hydrogen and nitrogen is C_6H_7N . The molar mass of the compound is 186.29g/mol

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~~The empirical formula of a compound
containing carbon ...~~

Carbon Compound Study Guide Answer

Covalently bonded compound that
contains carbon, excluding carbonates and
oxides. Functional Group Portion of a
molecule that is active in a chemical
reaction and that determines the properties

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of many organic compounds. Biology
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Answer to: What is the name of a
compound with 10 carbons and two rings?

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~~Answer Key~~ ... most of them are large molecules of covalently bonded carbon atoms. ...

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Covalently bonded compound that
contains carbon, excluding carbonates and
oxides. Functional Group Portion of a

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~~Answer Key~~ molecule that is active in a chemical reaction and that determines the properties of many organic compounds. Biology Section 3-1 Review: Carbon Compounds Flashcards ...

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and techniques organic chemistry organic
chemistry is the study of carbon

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~~Answer Key~~ compounds that always contain carbon
and it is limited to other
elements compounds obtained from plants
and animals ...

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? Carbon always forms covalent bond. ?

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The bond formed by sharing of electron pair between two atoms are known as covalent atoms. • Noble gas configuration of Carbon ? Carbon is tetravalent, it does not form ionic bond by either losing four electrons (C^{4+}) or by gaining four electrons (C^{4-}). It is difficult to hold four extra electron and would require large

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~~Answer Key~~ amount of energy to remove four electrons.

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Key Answer Key

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Organic Compounds Answers Description
Carbon is the key in organic chemistry
there is an atom named carbon carbon is
one unique atom given the fact that it can
form very stable molecules that range in
size the study of chemicals is a very

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Answer. (a) Carbon forms large number of compounds since carbon is small in size and can form stable covalent bonds

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(catenation) and it shows tetravalency. (b)
Air holes of gas burner are made open
(adjusted) so that air can pass through,
which is needed for complete combustion,
so that heated vessels do not get
blackened.

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