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in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written  $A \rightarrow B$ , where the direction of the arrow shows the direction in which the electron pair is donated.

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MODULE 1: FUNDAMENTALS IN

CHEMISTRY 1. (a)(i) Define the term

bond energy [1. mark] The bond energy

represents: The quantity of energy evolved

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when the bond is formed. Also, the bond energy is the amount of energy absorbed when the bond is broken. (ii) The bond lengths and bond energies of carbon-carbon single, double and triple bonds are listed below.

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