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Curriculum of Chemical Engineering: Texas A\0026M University
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Normality (N)—(gram equivalents of solute)/(liters of solution).

Oxidation—the loss of electrons by an atom or group of atoms.

pH—the negative logarithm (base 10) of the hydrogen ion

concentration in gram ions per liter; Reduction—the gain of electrons by an atom or group of atoms. Solubility product (S.P. or K.

ChemE

Book: Basic Principles and Calculations in Chemical Engineering

(8th Edition) Author: David M. Himmelblau and James B. Riggs

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